

Review: Chapters 24, 25, sulfonation from 26, and 27 through Monday the 11th

Mechanism:

nucleophilic substitution: Gabriel synthesis, α -halocarbonyl or carboxyl + nucleophile

nucleophilic addition and addition/elimination: lithium enolate of ketone or ester + aldehyde or ketone, Claisen & Dieckmann, aldol, HCN addition, lactone formation

decarboxylation: β -ketoacids and β -dicarboxylic acids

E₂ elimination: Hoffmann elimination, exhaustive methylation of amine followed by Ag₂O

rearrangements: Hoffmann, Curtius, and Schmidt degradations; benzidine, Wolff and Baeyer-Villiger rearrangements

synthesis of aryl diazonium ions: aryl amine + NaNO₂ (sodium nitrite) + acid

syn additions: peracid epoxidation (note opening of epoxide gives trans diol), OsO₄ (cis diol), Sharpless epoxidation

oxidative cleavage: HIO₄; forms a cyclic intermediate and requires cis stereochemistry for diol substrate

free radical reactions: acyloin condensation, metallic sodium + diester; Gomberg-Bachman, biphenyls

Wittig Reaction: alpha-bromoester + Ph₃P + methoxide, then react with ketone or aldehyde

Acid/base chemistry

amines are weak bases; resonance effects on basicity

difunctional compounds: nitroalkanes, dicarboxylic acids, β -ketoesters and β -diketones; review earlier acid/base chemistry

Resolution of enantiomeric carboxylic acids by making diastereomeric salts with optically active amines followed by recrystallization

Phase transfer catalysis

Nomenclature: amines, difunctional compounds

Reactions:

carbon-carbon bond forming reactions:

Mannich reaction: aldehyde or ketone + HCHO + amine hydrochloride

benzidine rearrangement to make biphenyls

Henry reaction: nitroalkane + aldehyde + hydroxide to make nitro alkenes

Wolff rearrangement: acid chloride + CH_2N_2 followed by rearrangement of alpha-diazoketone to ketene

Sandmeyer reaction with CuCN to make aryl nitriles

acyloin condensation (can also make 8-20 membered ring compounds)

aldol condensation

Claisen and Dieckmann condensations (make 5 and 6 membered ring compounds)

Wittig reaction with α -bromoester and Ph_3P to make alpha beta unsaturated esters

rearrangement reactions:

Hoffmann, Curtius, Schmidt, Wolff, benzidine, Baeyer-Villiger

degradation reactions:

Hoffmann, Curtius, Schmidt: carboxylic acid to amine with loss of carboxyl carbon

substitution reactions

Gabriel synthesis, sulfonation of aromatic ring ($\text{H}_2\text{SO}_4 + \text{SO}_3$), substitution of diazonium group of aryl diazonium salt for OH, OR, I, Br, Cl, CN, F, NO_2 , or H, α -bromoketone or ester + hydroxide

eliminations: Hoffmann (exhaustive methylation, Ag_2O , anti stereochemistry, least substituted alkene);

Cope (amine oxide, syn stereochemistry): both convert amines to alkenes

oxidations:

Sharpless epoxidation of allylic alcohol ($\text{t-BuOOH} + \text{Ti}(\text{O-iPr})_4 + \text{DET}$) followed by LAH reduction

peracid epoxidation of alkene (metachloroperbenzoic acid) followed by hydrolytic ring opening

Osmium tetroxide + N-methylmorpholine N-oxide, potassium permanganate (alkenes to syn-diols)

selenium dioxide, SeO_2 (ketones to α -diketones)

periodic acid, HIO_4 (syn-diols to two aldehydes, α -ketols to aldehyde+carboxylic acid)

Baeyer-Villiger: ketone plus $\text{CF}_3\text{CO}_3\text{H}$ (trifluoroperacetic acid) to make ester or lactone

reductions:

reduction of nitrogen compounds: catalytic reduction with H_2 , LiAlH_4 , NaBH_4 , NaH_3BCN